# SMITH, GAMBRELL & RUSSELL, LLP

ATTORNEYS AT LAW

SUITE 800

1850 M STREET, N.W.

Washington, D.C. 20036

TELEPHONE
(202) 659-2811
FACSIMILE
(202) 659-1462
WEBSITE
www.sgrlaw.com

May 19, 1999

ATLANTA OFFICE

SUITE 3100, PROMENADE II
1230 PEACHTREE STREET, N.E.
ATLANTA, GEORGIA 30309-3592
(404) 815-3500
FACSIMILE (404) 815-3509



Assistant Commissioner for Patents Box PATENT APPLICATION Washington, D.C. 20231

PRACTICING AS THE

BEVERIDGE, DEGRANDI, WEILACHER & YOUNG

INTELLECTUAL PROPERTY GROUP

#### PATENT APPLICATION TRANSMITTAL LETTER

Re:

Inventor: ASLANOVA

Title: METHOD AND APPARATUS FOR PRODUCING BASALTIC FIBERS

Attorney Docket No.: 33611YW002

Sir:

Transmitted herewith for filing are the following:

- New patent application including 13 pages of text, a signed Declaration, and one sheet of informal drawings on plain paper;
- A copy of PCT publication WO 98/22401;
- Exhibit A which is a literal English translation of International Application PCT/RU97/00355, of which the present application is a continuation.

Respectfully submitted,

Beveridge Degrand's etal

02-4200

Richard G. Young

Registration No. 20628

RGY:

5

#### METHOD AND APPARATUS FOR PRODUCING BASALTIC FIBERS

#### Field of the invention

The present invention relates to a method and apparatus for manufacturing mineral fibers from natural materials of the basalt group (basalts, andesitobasalts, andesites, gabbro etc.) which can be used in the construction, textile and chemical industries.

### **Reference to Related Application**

This is a continuation of International Application PCT/RU97/00355 filed under the Patent Cooperation Treaty on November 18, 1997, the entirety of which is incorporated by reference into this specification.

#### **Background of the Invention**

There are three main types of rock composition of the basalt group. The first type: rock composition enriched with oxides of iron and titanium ( $\sim$ 70% of Fe<sub>2</sub>O<sub>3</sub> and 20% of TiO<sub>2</sub>). The second type: basalt rocks enriched with oxides of aluminum and silicon ( $\sim$ ±25% of Al<sub>2</sub>O<sub>3</sub> and 55% of SiO<sub>2</sub>). The third type: basalt rocks enriched with oxides of magnesium, calcium and iron ( $\sim$ 12% of MgO, 20% of CaO, 10% of Fe<sub>2</sub>O<sub>3</sub>).

All these compositions are intended for basaltic fiber manufacture. However, to obtain temperature and chemical resistant fiber of high quality, the basalt rock composition is limited by the content of oxides. For example, in order to produce basaltic fibers, a glass is known containing the oxides SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>FeO, MnO, CaO, MgO, K<sub>2</sub>O, Na<sub>2</sub>O, SO<sub>3</sub>, P<sub>2</sub>O<sub>5</sub>, Se<sub>2</sub>O<sub>3</sub>, ZnO with the relations of constituents Al<sub>2</sub>O<sub>3</sub>/CaO+MgO <2.0, ensuring increased acid

5

resistance and temperature range of manufacture (Russian patent 2039019, class CO3C13/02, 1995).

However the known composition of glass makes it possible to obtain high content of  $A1_2O_3$  only in the specified range. This limits the use of basalts of other types and with other relations of oxides, and it eliminates the possibility of manufacturing from them a good acid and alkali resistant fiber of high heat stability.

The fiber manufacture from a glass mass of each individual composition requires certain production know-how. The closest method to the proposed one in its technical essence and the obtained result is a method for producing basaltic fibers which includes charging, melting of basalt in the interior of a furnace, feeding the melt into a feeder and stabilizing the glass mass, manufacture of fiber through a feeding unit, pulling the fiber through spinnerets, oiling the fiber, and reeling the fiber onto bobbins (Russian patent 2039715, class CO3B37/02, 1995).

The closest device to the proposed one is a device for producing basaltic fibers which includes a basalt weigher, a melting furnace, a feeder with discharging devices, feeding units, spinnerets, mechanisms for applying oil, and mechanisms for reeling the fibers up onto bobbins (Russian patent 2039715, class C03B37/02,1995).

Disadvantages of the known method and device are: not very high quality of the fiber at a low percent of manufacture, the complexity of the production process because of the necessity of the preparation of basalt rock, the necessity of a high temperature range in the melting furnace, and a long cycle of glass mass stabilization that involves the possibility of its crystallization and hence the vitrification on the surface of spinnerets.

20

25

30

5

## **Summary of the Invention**

An object of the invention is to provide a method and a device for obtaining corrosion resistant, heat stable continuous fibers out of basalt rocks of numerous compositions and to simplify the technology and the apparatus for manufacturing such fibers.

The technical result of the realization of the proposed method and device is to widen the technical possibilities of using basalt rocks of a wide range with a reduced process cycle, to increase the stability of the process, and to improve the strength, corrosion resistance and heat stability of the fiber.

The technical result is obtained in a method for producing basaltic fibers which includes the steps of preheating the basalt, charging the preheated basalt into a melting furnace, melting the basalt to form a glass mass, keeping the melted glass mass in a stabilizing section of the melting furnace until it reaches the fiber manufacture temperature, further stabilizing the glass mass in a feeder, pulling the fiber from spinnerets, oiling the fiber, and reeling it up onto bobbins. Stabilization in the feeder is carried out to obtain a glass mass composition with the relation of basic constituents

$$\frac{\text{Al}_2\text{O}_3 + \text{SiO}_2}{\text{CaO} + \text{MgO}} \ge 3 \qquad \qquad \frac{\text{FeO}}{\text{Fe}_2\text{O}_3} \ge 0.5$$

$$\frac{2Al_2O_3 + SiO2}{2 Fe_2O_3 + FeO + CaO + MgO + K_2O + Na_2O} > 0.5.$$

The technical result is best attained when:

- before charging into the furnace, the basalt is heated to 150 900°C;
- the fiber manufacture temperature is maintained equal to t melt +(50 250 °C) where t melt is

30

5

a temperature range of basalt melting;

- stabilizing of the glass mass in the feeder is carried out at a temperature of 1250 - 1450°C.

The technical result is obtained in a device for producing basaltic fibers, which includes a basalt weigher, a melting furnace, a feeder with discharging devices, feeding units, spinnerets, mechanisms for applying oil, and mechanisms for reeling the fibers up onto bobbins. According to the invention, a heat exchanger connects the basalt weigher with a firing space of the melting furnace, and the melting furnace has a stabilizing section for stabilizing the melted glass mass. The stabilizing section is connected with the feeder. The best technical result is attained when the height of the stabilizing section is 0.4 - 0.6 of the height of the inner space of the furnace. A heat exchanger preliminarily heats the basalt before it is charged into the furnace. The glass mass is stabilized to obtain glass mass composition with the relation of basic constituents

$$\frac{\text{Al}_2\text{O}_3 + \text{SiO}_2}{\text{CaO} + \text{MgO}} \ge 3 \qquad \frac{\text{FeO}}{\text{Fe}_2\text{O}_3} \ge 0.5$$

$$\frac{2Al_2O_3 + SiO2}{2 Fe_2O_3 + FeO + CaO + MgO + K_2O + Na_2O} > 0.5,$$

making it possible to remove crystal water, gas bubbles and foam, to stabilize the volume of the glass mass to obtain an even and smooth surface, and to ensure the stability of the temperature range and viscosity which is essential for fiber manufacture. The presence of a heat exchanger in the weigher on simultaneous charging ensures uniform heating throughout the volume of basalt by the reduction of hot air flowing from the firing space of the melting furnace, enabling the utilization of waste gases and the reduction of fuel consumption. The stabilizing section which has a height of 0.4 - 0.6 of the height of the furnace interior space contributes to stabilizing the melt in volume at the furnace exit with a specified temperature. The stabilizing section height is

20

5

determined by the melt height as the temperature decreases, and the possible exit of gases and foam.

The invention utilizes new combinations of technical features which satisfy the "novelty" and "inventive step" criteria. The utilization of the invention serves to increase the productivity of the melting furnace, and simultaneously decrease fuel and power consumption. This fact establishes that the proposed method and device satisfy the "industrial applicability" criterion.

# **Brief Description of the Drawings**

Fig. 1 illustrates an apparatus for producing basaltic fibers, using the process of the invention.

# **Detailed Description**

The apparatus is a plant, which has a weigher 1 for basalt 2 charging and a heat exchanger 3, connected with a firing space 4 of the melting furnace 5. The melting furnace 5 has a stabilizing section 6 in which the melted glass mass becomes stable in volume when reaching the temperature of fiber manufacture. The melting furnace 5 and stabilizing section 6 have heating systems 7. The stabilizing section 6 of the melting furnace 5 is connected to a feeder 8 where the melt becomes stable till averaging the mass and ensuring the relation of constituents in the composition. The feeder 8 has discharging devices 9 and feeding units 10 delivering the melt into spinnerets 11 through which the basaltic fibers 12 are pulled. Then the fibers 12 are supplied to oiling mechanisms 13 and mechanisms 14 for reeling the fibers onto bobbins.

The basalt compositions used in practicing the invention are given in Tables 1-4.

According to the method, basalt rocks are first cleansed of impurities, powdered, and

5

delivered through the weigher 1 into the melting furnace 5. The weigher 1 is connected with a heat exchanger 3 where basalts 2 are heated to a temperature 150 - 900 °C by hot air coming from the firing space 4 of the furnace 5. The heated basalts 2 enter the melting furnace 5 where they melt at a temperature of 1450 °C  $\pm 50$  °C until a glass mass melt is formed. Then the glass mass melt enters the stabilizing section 6 of the melting furnace 5. The limited height of the stabilizing section 6 ensures the stabilization and temperature reduction to a temperature of fiber manufactures which is t <sup>melt</sup> + (50 - 250 °C). In the stabilizing section 6, gas bubbles and foam are expelled and the surface becomes smooth and even. The melting furnace 5 and its stabilizing section 6 have heating systems 7. Out of the stabilizing section 6, a partially stabilized melt of glass mass enters the feeder 8 for averaging and obtaining the composition necessary for fiber manufacture. The feeder 8 also has heating systems 7 to maintain a temperature range of fiber manufacture (1350 - 1450 °C) and a viscosity of 60 - 240 Pa/s.

Example of glass mass compositions and production process conditions of fiber manufacture are presented in Table 5 and 6.

Out of the feeder (8) the melt of glass mass is delivered by a stream feeding unit(9), through feeding units (10) to spinnerets (11). Elementary threads of the fiber (12) are pulled from the spinnerets, oiled by mechanisms (13), and reeled up onto bobbins (14).

Physico-mechanical properties of the basalt fibers are shown in Table 7.

As will be seen from the Table 7, the method and apparatus according to the invention make it possible to obtain high-strength, corrosion resistant, heat stable continuous fiber out of basalt rocks of numerous compositions, and to simplify the technology of its manufacture.

Composition number				Co	mpositio	n of rock	base			
	Na	Mg	Al	Si	K	Ca	Ti	Mn	Fe	P
1.	4,567	0,232	11,537	32,932	2,426	1,428	12,771	0,240	33,968	-
2.	0,415	13,552	1,153	51,318	0,184	21,752	1,320	0,309	9,999	•
3.	6,573	0,358	20,340	60,648	4,873	2,088	1,506	0,001	2,689	0,326
4.	3,513	4,067	11,235	44,778	2,670	7,883	5,325	0,474	19,651	0,454
5.	5,744	0,465	19,541	56,221	4,503	3,924	2,889	0,180	5,642	0,890

Table 2

Composition number				Comp	osition o	f large inc	lusions			
	Na	Mg	Al	Si	K	Ca	Ti	Mn	Fe	P
1.	5,420	0,352	26,824	54,104	0,461	10,875	0,330	0,061	1,552	0,00
2.	6,672	0,000	20,207	64,108	6,410	1,540	0,300	0,024	0,489	0,248
3.	1,425	13,499	2,304	50,003	0,166	19,882	1,917	0,216	10,279	0,871
4.	0,984	0,685	24,053	56,550	4,568	8,310	2,847	0,031	1,992	0,00
5.	4,160	1,859	17,890	58,470	4,688	5,817	0,497	0,245	6,378	0,00

Table 3

Composition number				Compo	sition of	small in	clusions			
	Na	Mg	Al	Si	K	Ca	Ti	Mn	Fe	P
1.	5,775	0,413	18,112	63,813	8,139	1,459	0,132	0,000	2,156	0,000
2.	11,614	2,263	22,164	55,601	0,260	2,243	0,159	0,098	3,819	1,776
3.	0,422	1,364	0,817	0,830	0,086	0,214	23,541	1,226	71,502	0,000
4.	0,371	2,138	1,035	0,627	0,095	0,060	20,530	0,796	72,217	0,134
5.	0,727	12,683	1,364	49,475	0,187	20,085	2,023	0,250	13,121	0,087

Table 4

Composition number				Average (	composit	ion of sta	rting bas	alt		
	Na	Mg	Al	Si	K	Ca	Ti	Mn	Fe	P
1	6,325	1,970	17,833	55,903	4,553	4,672	1,532	0,160	6,977	0,000
2.	5,058	7,932	14,127	46,164	2,320	4,697	1,343	0,396	16,461	1,512
3.	5,877	2,773	17,493	53,716	8,923	4,867	1,299	0,098	8,276	1,680
4.	4,587	3,187	17,660	52,501	3,927	5,515	1,701	0,155	8,541	1,953
5.	4,404	3,470	16,824	51,606	2,810	7,681	1,852	0,185	9,223	2,944

Com- po- sition num- ber					Glass	mass co	mposit	ion for	fibre p	ulling			
	Na	Mig	Al	Si	K	Ca	Ti	Mn	Fe	P	Al <sub>2</sub> O <sub>3</sub>	FeO	2Al <sub>2</sub> O <sub>3</sub> +
	3.00				·						+SiO <sub>2</sub> CaO+ MgO	Fe <sub>2</sub> O <sub>3</sub>	SiO <sub>2</sub> 2Fe <sub>2</sub> O <sub>3</sub> + + FeO + + CaO + + MgO + + K <sub>2</sub> O + + Na <sub>2</sub> O
1.	2,00	10,58	11,82	50,42	0,52	8,84	1,04	8,18	12,25	0,21	3,2	3,34	2,0
2.	2,34	5,47	12,58	49,03	0,66	9,53	2,85	0,32	14,03	0,30	4,11	2,62	2,06
3.	3,88	4,65	16,75	50,61	1,0	9,07	1,81	0,18 -	10,26	0,40	1 4,9	0,54	2,37
4.	2,93	5,99	14,89	50,15	0,34	3,82	2,04	0,22	12,05	1,98	4,38	1,52	2,37
5.	4,75	3,54	15,33	49,66	3,10	6,56	2,84	0,21	12,05	1,98	6,44	1,62	2,39

Table 6

Composition number	Point of crystalization upon limit	Fibre diametre	Heat range of fibre manufacture	Viscosity range at T fmhr
	T culp °C	mcm	T fmhr °C	Pa C
1.	1290	8,4 – 12	1360 – 1400	104 – 62
2.	1275	7,0 – 13	1380 – 1440	112 – 64
3.	1240	7,0 – 11	1370 – 1450	188 – 64
4.	1250	7,0 – 12	1350 – 1440	235 – 96
5.	1245	7,0 – 12	1350 - 1430	235 – 104

Table 7

Composition number		Strengti	n and chemica	l resistance of	fibre	
	Fibre diametre	Tensile strength	Chemical	resistance in %	6 after three-h	our boiling
	mcm	MPa	H <sub>2</sub> O	Na	OH ·	HCI
				0,5H	2H	2H
1.	10,2	2400	99,3	92,6	85,3	75,9
2.	10,0	3110	99,4	97,5	94,0	80,6
3.	9,0	2240	99,5	98,2	95,2	91,0
4.	9,5	3050	99,4	97,6	96,8	90,1
5.	9,5	3100	99,4	94,1	92,5	83,5

## I Claim:

1. A method for producing basalt fibers, comprising the steps of:

preheating basalt;

charging the preheated basalt into a melting furnace;

heating the basalt in said furnace to form a glass mass;

stabilizing the glass mass in a stabilizing section of the melting furnace until it reaches a fiber manufacturing temperature;

introducing the stabilized glass mass into a feeder;

further stabilizing the glass mass in the feeder to obtain a glass mass having the composition

$$\frac{Al_2O_3 + SiO_2}{CaO + MgO} \ge 3 \qquad \qquad \frac{FeO}{Fe_2O_3} \qquad \ge 0.5$$

$$\frac{2Al_2O_3 + SiO2}{2 Fe_2O_3 + FeO + CaO + MgO + K_2O + Na_2O} > 0.5$$

forming fibers by pulling the further stabilized glass mass from spinnerets which receive glass from the feeder.

- 2. A method according to claim 1 wherein the preheating step heats the basalt to a temperature of 150 900 °C.
- 3. A method according to claim 1 wherein the temperature of the glass mass from which the fibers are pulled is  $t^{melt} + (50 250 \, ^{\circ}\text{C})$ , where  $t^{melt}$  is the basalt melting temperature.

- 4. A method according to claim 3 wherein the preheating step heats the basalt to a temperature of 150 900 °C.
- 5. A method according to claim 1, wherein the glass mass is stabilized in the feeder at a temperature of 1250 1450 °C.
- $\dot{6}$ . A method according to claim 5 wherein the preheating step heats the basalt to a temperature of 150 900  $^{\circ}$ C.
  - 7. A method according to claim 6 wherein the temperature of the glass mass from which the fibers are pulled is  $t^{\text{melt}} + (50 250 \,^{\circ}\text{C})$ , where  $t^{\text{melt}}$  is the basalt melting temperature.
  - 8. Apparatus for producing basaltic fibers, comprisinga basalt weigher;a melting furnace having a firing space and a stabilizing section;
    - a heat exchanger connecting the basalt weigher to the firing space for preheating basalt which is charged into the melting furnace;
    - a feeder which receives molten glass from the melting furnace, said feeder being connected by the stabilizing section to the firing space; spinnerets which receive molten glass from the feeder; and

mechanisms which pull fibers from the spinnerets.

- 9. Apparatus according to claim 8 wherein the stabilizing section has a height which is 0.4
   0.6 of the height of the height of the firing space.
- 10. Apparatus according to claim 9 wherein the heat exchanger is operable to preheat the basalt to a temperature of 150-900°C.
- 11. Apparatus according to claim 9 including means for heating the glass mass from which the fibers are pulled to a temperature of t <sup>melt</sup> +(50 250 °C), where t <sup>melt</sup> is the basalt melting temperature.
  - 12. Apparatus according to claim 9 including means for maintaining glass mass at a stabilizing temperature which is 1250-1450°C.
  - 13. Apparatus according to claim 8 wherein the heat exchanger is operable to preheat the basalt to a temperature of 150-900°C.
  - 14. Apparatus according to claim 13 including means for heating the glass mass from which the fibers are pulled to a temperature of t  $^{\text{melt}}$  +(50 250  $^{\circ}$ C), where t  $^{\text{melt}}$  is the basalt melting temperature.
  - 15. Apparatus according to claim 13 including means for maintaining glass mass at a stabilizing temperature which is 1250-1450°C.

- 16. Apparatus according to claim 8 including means for heating the glass mass from which the fibers are pulled to a temperature of t <sup>melt</sup> +(50 250 °C), where t <sup>melt</sup> is the basalt melting temperature.
- 17. Apparatus according to claim 8 including means for maintaining glass mass at a stabilizing temperature which is 1250-1450°C.
- 18. Apparatus according to claim 17 including means for heating the glass mass from which the fibers are pulled to a temperature of t <sup>melt</sup> +(50 250 °C), where t <sup>melt</sup> is the basalt melting temperature

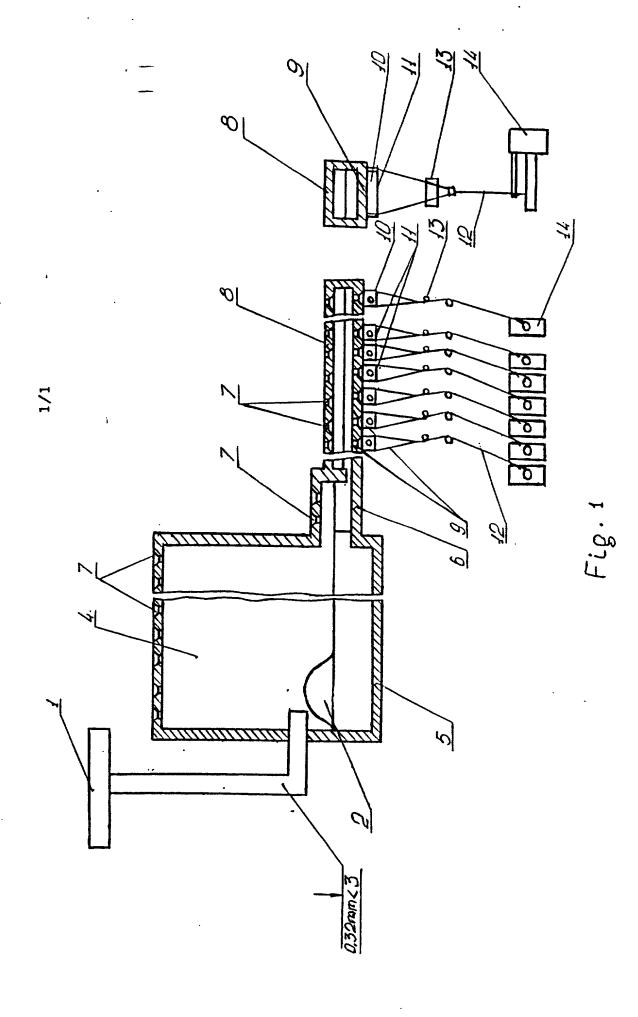
#### **Abstract of The Disclosure**

Mineral fibers made of natural basalt materials are produced by preheating basalt to a temperature of between 150 and 900 °C, loading the preheated basalt into a melting furnace, melting the basalt to form a glass mass, moving the melted glass mass through a stabilization zone of the melting furnace until a fiber production temperature of t <sup>melt</sup> + (50 - 250 ° C) is reached, further stabilizing the glass mass in a feeder at a temperature of 1250 to1450°C to obtain a glass mass having the composition

$$\frac{\text{Al}_2\text{O}_3 + \text{SiO}_2}{\text{CaO} + \text{MgO}} \ge 3 \qquad \frac{\text{FeO}}{\text{Fe}_2\text{O}_3} \ge 0.5$$

$$\frac{2Al_2O_3 + SiO2}{2 Fe_2O_3 + FeO + CaO + MgO + K_2O + Na_2O} > 0.5$$

The further stabilized glass mass is introduced to a feeding unit, and the fibers are drawn through dies, oiled, and wound onto reels. The apparatus includes a melting furnace and a basalt dosing unit which includes a heat exchanger connected to the firebox of the furnace. The furnace has a firing space where the basalt is melted to form a glass mass, and a stabilization zone where the mass glass is stabilized. A feeder receives molten glass from the stabilization zone and supplies the glass to dies from which the fibers are drawn. Mechanisms are provided for lubricating the fibers and winding them onto reels. The invention shortens the industrial cycle and increases the fiber resistance and thermal endurance.



BEVERIDGE DEGRA

Ø003:008

# Declaration and Power of Attorney United States Patent Application

UNITED STATES Potents and Design Paterus Sale & Joint Inventors
Convention & Non-convention
PCT & Non-PCT This form cannot be newaded, aftered or changed after it is signed (For use only for inventors who understand the English language.)

As a below named inventor, I hereby declare that: My residence, past office address and citizenship are as stated below next to my hame. I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled.

METHOD FOR PRODUCING BASALTIC FIBRES

1 IJ 

High High High I

AND DEVICE FOR REALISING THE SAME

	is attached her was filed as U	eto. S. Application No	on	and (If
.•	applicable) was filed as P	is amended on	CT/RU97/00355 on Nove	mber 18, 1997 and (if applicable) was
I acknowledge	the duty to disclose infor	rents of the above-klentified specification, matter which is material to patentability a	is defined in Title 37, Code of Fed	leral Regulations, §1.56.
certificate, or {	365(a) of any PCT inter	national application which designated at le application for patent or inventor's certifi	ast one country other than the Uni	ad PCT application(s) for patent or inventor's ited States of America listed in this Declaration ion having a filing due before that of the
Foreign/i	PCT Application No.	Country	Filing Date	Priority Claimed? (yes/no)
96 122192 PCT/RU97/0	10355	Russia WIPO (International Bufeau)	19 November, 1996 18 November, 1997	yes yes
disclose inform	acion which is material !	o patentability as defined in Title 37, Code	e of Federal Regulations, \$1.56 wi	i States Code, §112, I acknowledge the duty to thich became available between the filling date of
	etion and the national or	PCT international filing date of this appli	cation:	ntea/pending/upandonéd/;
I hereby appoint W. Collier (42 (29405), Michael G. Yo	Application No.  In the following atterneys (429), loseph A. DeGran acl A. Mukuch (32263), ung (20628),	Filing Date  to prosecute this application and to transcible (17446), Thomas L. Evans (35805), Ca. William F. Rauchholz (34701), Dennis C.	Stonus (gale)  Stonus (gale)  act all business in the Patent and Trolyn A. Pavorito (39183), Herber Rodgers (32936), Charles L. Wan	rademark Office connected therewith: Steven M. Hanegan (25682), J. Rogers Lunaford, III oner, II (32330), Robert G. Weilacher (20531),
I hereby appoint W. Collier (42 (29405), Michael G. Yo Send all corres 800), Washington	Application No.  In the following atterneys A29), Joseph A. DeGrah aci A. Mukuch (32263), sung (20628), spondence to: Smith, Gar on, D.C. 20036. All fac	Filing Date  to prosecute this application and to transal (17446), Thomas L. Evans (35805), Ca. William F. Rauchholz (34701), Dennis C. Inbroll & Rusself, LLP, Beveridge, DcGrasimiles may be sent to (302) 659-1462.	stones (pale)  Stones (pale)  Let all business in the Patent and Trolyn A. Pavorito (39183). Herber Rodgers (32936), Charles L. Wan  Indi, Weitscher & Young Intellect  liters all phone calls (a (202) 659-1	rademark Office connected therewith: Steven M. Hanegan (25682), J. Rogers Lunatond, III mer, II (32320), Robert G. Weslacher (20531), asl Property Group, 1850 M Street, N.W. (Strietting)
I hereby appoint W. Collier (42 (29405), Michard G. Yo Send all corres 800), Washings I hereby declar be true; and fur or both, under	Application No.  In the following autorney: 429), Joseph A. DeGran acl A. Mukuch (32263), sing (20628), spondence to: Smith, Gar on, D.C. 20036. All fac	Filing Date  to prosecute this application and to transal (17446), Thomas L. Evans (38805), Ca. William F. Rauchholz (34701), Dennis C. abrell & Russell, Ll.P. Beveridge, DoGra similes may be sent to (302) 639-1462. Dute herein of my own knowledge are true at the work made with the knowledge that will on the United States Code and that such we	Atomic foote statements and the like statements and the statements and the like statements and the lik	rademark Office connected therewith: Steven M. Hanegan (25682), J. Rogers Lunaford, III (123320), Robert G. Weilacher (20531), and Property Group, 1830 M Street, N.W. (Strictmanton and bettef are believed to made are punishable by fine or imprisonment.
I hereby appoint W. Collier (42 (29405), Michael G. You Send all corress 800), Washings I hereby declar be true; and furnithe application Full name of Residence (	Application No.  In the following autorneys A29), loseph A. DeGrah aci A. Mukuch (32263), Impondence to: Smith, Gar on, D.C. 20036. All fac te that all statements mac rither that these statement Section 1001 of Title 18 or any parent issued the of joint inventor: Lit- circ state, country)	Filing Date  to prosecute this application and to transa di (17446), Thomas L. Evans (35803), Ca. William F. Rauchholz (34701), Dennis C. Inhrelf & Russell, Ll.P. Beveridge, DeGra similes may be sent to (202) 639-1462. Due herein of my own knowledge are true at its were made with the knowledge that will of the United States Code and that such wroon.	cation:  Stoates (palle)  act all business in the Patient and Trollyn A. Pavorito (39183), Herber Rodgers (32936), Charles L. Wannet, Weitscher & Young Intellectures all phone calls to (202) 659-31 at that all statements made on infeful false statements and the like scillful false statements and the like scillful false statements may jeapant.  Citiz	Trademark Office connected therewith: Steven M. Hanegan (25682), J. Rogers Lunaford, III (12320), Robert G. Weslacher (20531), and Property Group, 1850 M Street, N.W. (Stirl Property Group, 1850 M